

## What is the molecular mass of NaOH?

## A. 39.996 g/mol

B. 29.996 g/mol

C. 19.996 g/mol

D. 09.996 g/mol



Chemistry MCQs by CSS MCQs

## Calculation of Molecular Mass of NaOH

Sodium hydroxide (NaOH) has 3 elements i.e Na, O and H

We know that

Na x 1 = 22.989 g/mol

 $O \times 1 = 15.999 \text{ g/mol}$ 

 $H \times 1 = 1.008 \text{ g/mol}$ 

By adding the molar masses of all the 3 three elements, we get 39.996 g/mol



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